## New Preparation of *cis*-Divinylcyclopropane

By MANFRED P. SCHNEIDER\* and JOCHEN REBELL

(Institut für Chemie, Universität Hohenheim, D-7000 Stuttgart-70, Emil-Wolff-Str. 14, Germany)

Summary cis-Divinylcyclopropane has been prepared by low temperature photolysis of cis- and trans- 3,5-divinyl-1-pyrazoline.

THE rôle of cis-divinylcyclopropane (1) in the Cope rearrangement<sup>1</sup> has been under much discussion. According to earlier reports (1) could not be isolated even at -40 °C.<sup>2</sup> In contrast, the preparation of (1) via a Wittig reaction from the corresponding dialdehyde has been reported recently.3

We report a convenient preparation of (1) by photolysis of the cis- and trans-3.5-divinyl-1-pyrazolines (2c and 2t), which were prepared by 1,3-dipolar cycloaddition of 3diazoprop-1-ene to buta-1,3-diene (0-4 °C, 4 days), isolated at 0 °C and purified by trap-trap distillation.<sup>4</sup>

Thermal decomposition of (2c and 2t) between 0 and 50 °C produces only trans-divinylcyclopropane (3) and cyclohepta-1,4-diene (4), as does direct photolysis at 0 °C. Compound (4) could have been derived either via Cope rearrangement from (1) which was formed initially or by direct ring closure of the diallylic 1,3-diradical (5) which may be formed.

We have studied the low-temperature photolysis of (2c and 2t). Solutions of (2c and 2t) (45:55 cis: trans; 20%)in CDCl<sub>3</sub>) were irradiated in Pyrex n.m.r. tubes either in an all quartz lens-Dewar-assembly at -45 °C (Philips HPK 125 W, Zeiss filter UG 1) or in the probe of a 100 MHz (Varian HA 100, Philips SP 1000 W lamp) n.m.r. spectrometer.†

Irradiation of the mixture (2) at -45 °C gives (1) and (3) with a trace of (4).<sup>†</sup> The F.T. n.m.r. spectrum of the products [high field protons of (1):  $\delta$  (CDCl<sub>3</sub>; Me<sub>4</sub>Si; 90

† Courtesy of Professor Dr. H. Fischer, Universität Zürich, Switzerland.

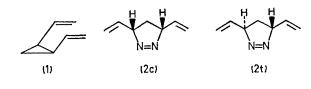
 $\ddagger$  The trace of (4) is derived from handling and is not visible in a control sample irradiated at -55 °C and stored at -78 °C.

<sup>1</sup> W. von E. Doering and W. R. Roth, Angew. Chem., 1963, 75, 27; Angew. Chem. Internat. Edn., 1963, 2, 115; E. Vogel, Angew Chem., 1962, 74, 829; Angew. Chem. Internat. Edn., 1963, 2, 1. W. v. E. Doering and W. R. Roth, Tetrahedron, 1963, 19, 715.

<sup>8</sup> J. M. Brown, B. T. Golding, and J. J. Stofko, jun., J.C.S. Chem. Comm., 1973, 319.

<sup>4</sup> M. Schneider and G. Mössinger, *Tetrahedron Letters*, 1974, 3081.

MHz) 1.76 (1- and 2-H), 1.22 (4-H), and 0.75 (3-H);  $J_{1,3}$ 6.0,  $J_{1,4}$  7.5, and  $J_{3,4}$  5.0 Hz] is unchanged after storage of the mixture for several weeks at -50 °C, showing that (1) does not rearrange to any large extent below 0 °C, as monitored by the resonances from (4) at  $\delta$  2.25 and 2.85. At +11 °C the rearrangement  $(1) \rightarrow (4)$  was carried out in the probe of the n.m.r. spectrometer with  $t_1$  ca. 25 min.<sup>3</sup>





After completion of the Cope rearrangement, the reaction mixture consisted of 53% (3) and 47% (4). The photochemical decomposition of (2c and 2t) give (1) and (3) exclusively. The n.m.r. data give no indication of direct formation of (4) by ring closure of a diallylic 1,3diradical (5).

We thank Dr. John M. Brown, University of Oxford, for stimulating discussions and for a spectrum of (1).

(Received, 7th February 1975; Com. 138.)